

SOIL FERTILITY, RATOON SUGARCANE YIELD, AND POST-HARVEST RESIDUES AS AFFECTED BY SURFACE APPLICATION OF LIME AND GYPSUM IN SOUTHEASTERN BRAZIL

FERTILIDADE DO SOLO, PRODUTIVIDADE DE CANA SOCA E RESÍDUOS PÓS-COLHEITA AFETADOS PELA APLICAÇÃO SUPERFICIAL DE CALCÁRIO E GESSO NA REGIÃO SUDESTE DO BRASIL

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ABSTRACT: The aim of this study was to evaluate soil fertility amendment, stalk yield, sugar, bagasse, and trash yield as affected by surface application of dolomitic limestone and gypsum in green sugarcane ratoon. A factorial arrangement of four dolomitic limestone rates (0, 900, 1800, and 3600 kg ha⁻¹) and two gypsum rates (0 and 1700 kg ha⁻¹) was used in the experiment. The experiment was performed from October 2006 to October 2007, starting in the rainy season after the 2nd ratoon harvest of the variety SP813250 in a commercial green sugarcane plantation of the São Luiz Sugar Mill (47°25'33" W; 21°59'46", 627 m of altitude), located in Pirassununga, state of São Paulo, in southeast Brazil. After 12 months, the experiment was harvested, technological measurements of stalk yield were made, and soil samples were taken and analyzed. Gypsum acts as a subsurface conditioner thus contributing to the benefits of surface limestone application, on soil acidity correction, to reach deeper layers, allowing the development of the root system of plants in greater depth. However, application of gypsum may lead to leaching of Mg and K from the topsoil layers. The higher rates of lime in surface application provided acidity amendment and, consequently, increased soil fertility in the soil profile, and increased sugarcane stalk, sugar, bagasse, and trash yield.

KEYWORDS: *Sacharum spp.* Soil chemical properties. Stalks. Sugar. Bagasse and trash.

INTRODUCTION

Soil acidity limits agricultural production in a considerable part of the world. In Brazil, approximately 70% of the land is composed of acid soils, which may reduce crop production potential by around 40%. Plants do not develop well in acid soils, above all because of Al toxicity and Ca deficiency (RAIJ, 2011).

To correct soil acidity, the materials used are commonly oxides, hydroxides, silicates, and carbonates. Limestone is the most widely used material (ALCARDE; RODELLA, 2003). Gypsum, the residue from production of triple superphosphate, and even mineral calcium sulfate, are used as soil conditioners since they have high solubility and mobility in the soil profile, without amending soil acidity. These two products lead to availability of the Ca²⁺ and SO₄²⁻ ions in the soil solution, enriching the subsoil layers with nutrients and reducing Al³⁺ saturation at soil depths (ALCARDE; RODELLA, 2003).

Recommendations for soil acidity amendments in sugarcane were developed in a

burned sugarcane harvest system. Currently, harvest of green sugarcane, without burning, is already a reality in 70% of the cane fields of São Paulo and has been growing significantly in most regions of Brazil. In sugarcane harvest without prior burning, a straw trash cover from 10 to 20 t ha⁻¹ year⁻¹ of dry matter remains on the soil surface (VITTI et al., 2008). The application of soil acidity amendments in this system where the straw remains on the soil has been carried out with little scientific foundation.

Studies on soil amendment in areas under the no-tillage system for grain production have shown that the effect of liming applied on the surface for amendment of subsoil layers varies with the application rate and particle size of the product, the form of application, the type of soil, the climatic conditions (especially water regime), the growing system, and the period of time since application (CAIRES et al., 2005; ALLEONI et al., 2005; SORATTO; CRUSCIOL, 2008a; ROSSATO et al. 2009), which leads to controversy regarding the effectiveness of this practice, particularly for acidity amendment of the subsoil.

It is known that soils after liming exhibit a fall in base saturation at the time of renovation of the sugarcane field and decreases in Ca, Mg, and K over the period of cuttings (MORELLI et al., 1992). Associated with the fall in soil fertility, there is a progressive decline in production, which, over the period of cuttings, will culminate in the need for renewal of the cane field, which is one of the most costly phases of the production system. Thus, it is necessary to verify if limestone and gypsum application over the cane straw can lead to soil acidity amendment in the subsoil, increasing the yield and longevity of the cane field.

In recent years, the burning of bagasse from sugarcane not only provides electric power for operation of the sugar mills, but the surplus electric power produced is sold to electric power distribution companies. Driven mainly by the increase in electric power prices in the free market, mills are gradually increasing their installed capacity and electric power production (UDOP, 2014). According to the National Report on Power Balance (BEN, 2014), the increase in electric power from sugarcane bagasse in 2013 was 19.2% compared to the previous year. Participation in the Brazilian energy matrix increased from 4.2% in 2012 to 4.9% in 2013. Moreover, bagasse and trash (post-harvest residues) are beginning to be used for production of ethanol from second generation or cellulosic ethanol. Therefore, with the introduction of this new source of income, parameters such as the sugarcane bagasse and trash should be considered together with the stalk and sugar yield in making decisions about which agricultural practices should be adopted.

Thus the aim of this study was to evaluate soil acidity amendment, and stalk, sugar, bagasse, and trash biomass yield as affected by different rates of surface application of dolomitic limestone and gypsum on green sugarcane ratoon.

MATERIAL AND METHODS

The experiment was performed from October 2006 to October 2007, starting in the rainy season after the 2nd ratoon harvest of the variety SP813250 in a commercial green sugarcane plantation of the São Luiz Sugar Mill (47°25'33" W; 21°59'46", 627 m of altitude), located in Pirassununga, state of São Paulo, in southeast Brazil. The climate is Aw (Tropical Savanna, by the Köppen classification) and the area is flat. Daily rainfall was registered during the experiment, and the results showed 102 mm in October, 120 mm in November, 175 mm in December, 275 mm in

January, 172 mm in February, 164 mm in March, 62 mm in April, 58 mm in May, 18 mm in June, 15 mm in July, 15 mm in August, 28 mm in September, and 84 mm in October, for a total of 1,288 mm.

The soil is classified as kaolinitic, thermic Typic Haplorthox, with a sandy loam texture. Immediately after the harvest of the second ratoon and before initiating the experiment, soil chemical and textural characteristics were determined. Thus, ten subsamples were taken from the experimental area between the ratoon rows and combined into one composite sample with respective results for the topsoil (0.00-0.20 m) and subsoil (0.20-0.40 m) as follows: pH of 4.8 and 4.7, soil organic matter of 23.6 and 16.8 g dm⁻³, P (resin) of 7.0 and 4.0 mg dm⁻³, Si of 5.4 and 5.1, and S of 6.0 and 4.0. Exchangeable K, Ca, Mg, and Al, CEC (cation exchange capacity), and H+Al (total acidity pH 7.0), for the topsoil (0.00-0.20 m) and subsoil (0.20-0.40 m), were 1.7 and 1.0, 17 and 12.5, 6.15 and 4.9, 1.3 and 2.8, 53.9 and 49.0, 29.3 and 30.6 mmolc dm⁻³, respectively. Base saturation for the topsoil (0.00-0.20 m) and subsoil (0.20-0.40 m) was 46.1 and 37.6 %, respectively. Sand, silt, and clay contents for the topsoil (0.00-0.20 m) and subsoil (0.20-0.40 m) were 693 and 677, 43 and 34, and 264 and 289 g kg⁻¹, respectively.

Soil pH was determined in 0.01 mol L⁻¹ CaCl₂ (RAIJ et al., 2001), SOM (soil organic matter) by the colorimetric method (RAIJ et al., 2001), P from extraction by ion exchange resin and determination by colorimetry (RAIJ et al., 2001), S by 0.5 mol L⁻¹ NH₄OAc in 0.25 mol L⁻¹ HOAc (VITTI, 1988), K from extraction by ion exchange resin and determination by flame photometry (RAIJ et al., 2001), Ca and Mg from extraction by ion exchange resin and determination by atomic absorption spectrometry (RAIJ et al., 2001), H+Al (total acidity at pH 7.0) from determination by a potentiometer in SMP-buffer solution (RAIJ et al., 2001), Al by 1 mol L⁻¹ KCl (RAIJ et al., 2001), and Si from extraction by 0.5 mol L⁻¹ acetic acid and determination by a spectrophotometer at 660 nm (KORNDÖRFER et al., 1999).

A randomized block experimental design was used in a factorial arrangement, with four replications. Treatments consisted of four dolomitic limestone rates (0; 900; 1,800; and 3,600 kg ha⁻¹) and two gypsum rates (0 and 1,700 kg ha⁻¹). Each plot consisted of eight sugarcane rows of 12 m length, spaced at 1.4 m.

Dolomitic limestone composition was 40% CaO, 12.0% MgO, and 85% effective calcium carbonate equivalence (ECCE). The composition of gypsum (CaSO₄·2H₂O), a by-product obtained from

a Brazilian phosphoric acid industry, was 20% Ca, 16% S, and a small residue of 0.1% P and F.

The dolomitic limestone rate (LR) was calculated so as to increase the base saturation in soil (0.00-0.40 m) to 60 % according to Spironello et al. (1996). For this purpose, the soil chemical characteristics determined in topsoil (0.00-0.20 m) were used, and the result obtained from the amount of dolomitic limestone is multiplied by two (SPIRONELLO et al., 1996) as follows in equation 1:

$$LR (kg \text{ ha}^{-1}) = \{[(BS_2 - BS_1) \cdot CEC] / (ECCE \cdot 10)\} \cdot 2 \quad [1]$$

BS_2 is the estimated base saturation (60%) and BS_1 is the base saturation measured in soil analysis, as in equation 2:

$$BS(\%) = [(Ca_{ex} + Mg_{ex} + K_{ex}) / CEC] \cdot 100 \quad [2]$$

Ca_{ex} , Mg_{ex} , and K_{ex} are basic exchangeable cations and CEC is the total cation exchange capacity, calculated according to equation 3:

$$CEC (mmolc \text{ dm}^{-3}) = Ca_{ex} + Mg_{ex} + K_{ex} + \text{total acidity at pH 7.0 (H + Al)} \quad [3]$$

The lime rates were defined as the control (0 kg ha^{-1}), half the rate recommended (900 kg ha^{-1}), the rate recommended (1,800 kg ha^{-1}), and double the rate recommended (3,600 kg ha^{-1}). Before application of the products, lime and gypsum were mixed, according to the treatments, and they were packed in plastic bags.

The gypsum rate (GR) was calculated through equation 4:

$$GR (kg \text{ ha}^{-1}) = 6CL \quad [4]$$

CL is the clay content (g kg^{-1}) in the 0.20 - 0.40 m soil layer.

Thus, products were applied over the sugarcane harvest residue layer (12.2 ton ha^{-1}) and without soil incorporation, according to the treatments. Topdressing fertilization on the sugarcane ratoon was performed with 70 kg ha^{-1} of N as ammonium nitrate and 80 kg ha^{-1} of K_2O as potassium chloride.

After 12 months, the experiment was harvested and soil samples were taken at the depths of 0.00-0.05, 0.05-0.10, 0.10-0.20, 0.20-0.40, and 0.40-0.60 m in all plots. Seven subsamples were taken at random from each plot between the rows and combined into one composite sample. Soil samples were chemically analyzed for pH, SOM, H+Al, and P, exchangeable Al, Ca, Mg, and K (RAIJ et al., 2001) and S (VITTI, 1988). Using the results of exchangeable bases and total acidity at pH 7.0 (H+Al), base saturation values were calculated (RAIJ et al., 2001).

Stalk yield at harvest was determined in the four central rows, with two rows of plants being

used for technological measurements (sugar concentration, purity, fiber, and reducing sugar). One meter was established at random in both plant rows; the stalks were collected, topped at apical bud height, defoliated, and then sent to the sugarcane technology laboratory, São Luiz Sugar Mill, to be processed according to the methodology defined in the Sucrose Content-Based Sugarcane Payment System, in accordance with Consecana's semiannual updates for the technological evaluations described in Fernandes (2011), considering the sugarcane total reducing sugars parameter. Using the results of stalk yield and sugar concentration, the sugar yield values were calculated, and using the results of fiber and stalk yield, bagasse values were calculated with 50% moisture. The quantity of trash (post-harvest residues) was calculated considering that 1 Mg of stalk produces 140 kg of trash (HASSUANI et al., 2005) and that 50% of post-harvest residues were collected from the soil surface.

Data were subjected to ANOVA using SAS (SAS Institute, 1997). Gypsum application means were compared by Fisher's protected LSD test at the 0.05 probability level ($p < 0.05$). Dolomitic limestone rates were analyzed through regression analysis, adopting the magnitude of the regression coefficients that were significant at the 0.05 probability level ($p < 0.05$) by the t test as the criterion for choosing the model.

RESULTS AND DISCUSSION

The limestone application rates increased the pH in $CaCl_2$ of the soil in a linear manner and reduced exchangeable Al and saturation by Al up to the 0.40-0.60 m layer and potential acidity (H + Al) up to the 0.20-0.40 m layer (Table 1). This greater concentration of ions resulting from the neutralization reaction at the higher limestone application rates allows the negative charges of the soil arising from the increase of pH to be filled, making the excess of cations available to the soil solution in a shorter period of time, which makes displacement of the alkaline front possible in the subsoil, confirming the affirmation of Caires et al. (2004) for the application of acidity amendments on the soil surface.

Physical movement of the limestone by root channels, biological microchannels (biopores) and planes/zones of weakness in the soil itself (AMARAL et al., 2004), has been indicated as the main factor that contributes to acidity amendment at soil depths. In addition, high amounts of rainfall in the months following limestone application may have contributed to amendment of the soil profile.

Table 1. Values of pH, H+Al (total acidity pH 7.0), exchangeable Al³⁺ (Al_{ex}), and m% (Al³⁺ saturation) of soil samples taken 12 months after dolomitic lime and gypsum surface application in green sugarcane and ANOVA significance at five soil depths in Pirassununga, São Paulo, Brazil.

Characteristic	pH		H+AL		Al _{ex}		m%	
Gypsum, kg ha ⁻¹	0 ^(a)	1,700	0	1,700	0	1,700	0	1,700
	(CaCl ₂)		-----mmol _c dm ⁻³ -----				%	
Lime, kg ha ⁻¹	<u>depth: 0.0–0.05 m</u>							
0	4.9	5.1	30.0	29.5	1.45	0.95	4.8	3.4
900	5.3	5.4	24.3	24.0	1.15	0.90	2.9	2.5
1,800	5.4	5.4	23.9	22.7	1.20	0.90	2.7	2.2
3,600	5.9	6.1	18.4	17.7	1.10	0.80	2.1	1.5
Mean	5.4a ^(b)	5.4a	24.3a	23.5a	1.23a	0.88b	3.2a	2.4b
Effect	L ^(c) **		L**		L**		L**	
ANOVA								
Lime (L)	<0.0001		<0.0001		0.0237		<0.0001	
Gypsum (P)	0.1254		0.1152		<0.0001		<0.0001	
L*P	0.1552		0.1584		0.1358		0.1297	
Lime, kg ha ⁻¹	<u>depth: 0.05–0.10 m</u>							
0	4.8	5.0	29.8	28.3	1.6	1.1	6.3	4.4
900	5.0	5.1	26.4	25.4	1.4	1.0	4.7	3.7
1,800	5.1	5.2	25.3	24.8	1.3	1.0	4.2	3.1
3,600	5.4	5.6	22.8	17.3	1.2	0.9	3.1	2.0
Mean	5.1a	5.2a	26.1a	23.9a	1.4a	1.0b	4.6a	3.3b
Effect	L**		L**		L**		L**	
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		<0.0001	
Gypsum (P)	0.1291		0.1007		<0.0001		<0.0001	
L*P	0.1468		0.1437		0.1246		0.1205	
Lime, kg ha ⁻¹	<u>depth: 0.10–0.20 m</u>							
0	4.9	5.0	27.3	26.7	1.7	1.0	6.9	4.0
900	5.0	5.0	26.5	26.1	1.4	1.0	5.4	3.8
1,800	5.0	5.1	26.0	25.4	1.3	0.9	4.8	3.2
3,600	5.2	5.4	24.5	23.2	1.3	0.8	4.4	2.7
Mean	5.0a	5.1a	26.1a	25.4a	1.4a	0.9b	5.4a	3.4b
Effect	L*		L**		L*		L*	
ANOVA								
Lime (L)	0.0249		<0.0001		<0.0001		<0.0001	
Gypsum (P)	0.1476		0.1099		<0.0001		<0.0001	
L*P	0.1504		0.1369		0.1327		0.1311	
Lime, kg ha ⁻¹	<u>depth: 0.20–0.40 m</u>							
0	4.6	4.8	30.3	30.2	2.9	2.0	14.3	9.8
900	4.8	4.8	29.0	29.0	2.6	1.8	11.7	8.2
1,800	4.8	4.8	28.6	28.5	2.5	1.5	10.7	6.4
3,600	4.9	4.9	28.2	26.7	2.3	1.4	9.6	5.7
Mean	4.8a	4.8a	29.0a	28.6a	2.6a	1.7b	11.6a	7.5b
Effect	L*		L*		L**		L*	
ANOVA								
Lime (L)	0.0444		0.0156		<0.0001		<0.0001	
Gypsum (P)	0.2529		0.0997		<0.0001		<0.0001	
L*P	0.2873		0.1349		0.1266		0.1084	

Lime, kg ha ⁻¹	depth: 0.40–0.60 m							
0	4.2	4.4	36.6	33.3	4.5	3.5	29.5	17.5
900	4.4	4.5	31.9	29.8	4.1	3.3	25.3	16.5
1,800	4.4	4.5	34.0	30.2	4.0	3.0	22.6	14.5
3,600	4.5	4.7	31.2	30.8	3.7	2.5	19.4	12.2
Mean	4.4a	4.5a	33.4a	31.0a	4.1a	3.1b	24.2a	15.2b
Effect	L*		ns		L**		L**	
ANOVA								
Lime (L)	0.0487		0.0827		<0.0001		<0.0001	
Gypsum (P)	0.2116		0.0722		<0.0001		<0.0001	
L*P	0.2331		0.1171		0.1305		0.1091	

* Significant at the 0.05 probability level, ** significant at the 0.01 probability level and ^{ns} represents no effect. ^(a) Gypsum treatments (0 and 1,700 kg ha⁻¹). ^(b) Values followed by the same letter in the rows, within each variable, are not significantly different at $P \leq 0.05$ according to the LSD test. ^(c) L represents linear effect by polynomial regression.

No effect was observed from application of gypsum on pH and H +Al. However, application of gypsum reduced the exchangeable Al contents and saturation by Al for all the layers. Some hypotheses, described by Raij (2008), have been suggested to explain the mechanism involved. One of them would be the release of OH⁻ by SO₄²⁻ through ligand exchange, with the formation of hydroxylated structures of aluminum, a mechanism called the “self-liming effect”. Aluminum precipitation, with the formation of minerals, has also been indicated. Another possibility would be the leaching of aluminum accompanying the gypsum, which may in part be favored by the formation of ionic pairs or compounds such as AlSO₄⁺, or of aluminum fluoride.

Limestone application increased the Ca and Mg contents and base saturation in a linear and positive manner in all the soil layers evaluated, regardless of the application of gypsum (Table 2). These results corroborate those found by Soratto and Crusciol (2008a), who verified an increase in the Ca and Mg contents up to 0.60 m in an Oxisol 12 months after surface application of limestone in

establishment of a no-tillage system (NTS). Displacement of the Ca-ligand and the Mg-ligand to subsoil layers occurs when the stability constant of the complex is greater than the stability constant of the calcium-soil and of the magnesium-soil (AMARAL et al., 2004). The displacement of Ca²⁺ and Mg²⁺ in the soil profile by limestone may be attributed to their high concentration, especially for Ca²⁺ in the surface layers, since it is well above the range considered as adequate, 4.0 to 7.0 mmol_c dm⁻³ (RAIJ et al., 1996). Limestone was effective in raising base saturation to levels recommended for sugarcane (60%) in the 0.0-0.05 m and 0.05-0.10 m layers for the lower rate (900 kg ha⁻¹) and rate recommended by Spironello et al. (1996) (1800 kg ha⁻¹). For the highest rate (3600 kg ha⁻¹), this effect was observed up to the 0.05-0.10 m layer. This shows that the limestone application rate has a direct effect on the depth where the effect of the soil amendment is present. In addition, the results indicate that application rates higher than those recommended by Spironello et al. (1996) were more effective in amendment of the 0.0-0.10 m layer after 12 months.

Table 2. Values of exchangeable Ca, Mg, and K, and base saturation of soil samples taken 12 months after dolomitic lime and gypsum surface application in green sugarcane, and ANOVA significance at five soil depths in Pirassununga, São Paulo, Brazil.

Characteristic	Ca _{ex}		Mg _{ex}		K _{ex}		BS	
Gypsum, kg ha ⁻¹	0 ^(a)	1,700	0	1,700	0	1,700	0	1,700
	-----mmol _c dm ⁻³ -----						%	
Lime, kg ha ⁻¹	<u>depth: 0.0–0.05 m</u>							
0	18.1	20.9	9.6	5.9	0.78	0.51	48.7	48.1
900	24.8	27.5	12.5	7.3	0.74	0.47	61.0	59.5
1,800	29.8	31.3	12.6	8.1	0.69	0.32	64.3	63.6
3,600	36.6	44.5	13.2	8.8	0.55	0.26	73.2	75.2
Mean	27.3b ^(b)	31.1a	12.0a	7.5b	0.68a	0.39b	61.8a	61.6a

Effect	L ^{(c)**}		L*		L**		L**	
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		<0.0001	
Gypsum (P)	<0.0001		<0.0001		<0.0001		0.1659	
L*P	0.1269		0.1325		0.1553		0.2084	
Lime, kg ha ⁻¹	<u>depth: 0.05–0.10 m</u>							
0	16.2	18.2	6.7	5.2	0.74	0.37	44.2	45.7
900	19.2	19.3	8.3	6.7	0.69	0.37	51.6	50.9
1,800	19.7	23.3	9.8	7.3	0.47	0.31	54.2	55.5
3,600	26.7	35.9	9.9	7.4	0.41	0.28	61.9	71.6
Mean	20.4b	24.2a	8.7a	6.7b	0.58a	0.33b	53.0a	55.9a
Effect	L*		L**		L**		L**	
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		<0.0001	
Gypsum (P)	<0.0001		<0.0001		<0.0001		0.1271	
L*P	0.1415		0.1257		0.1308		0.1533	
Lime, kg ha ⁻¹	<u>depth: 0.10–0.20 m</u>							
0	17.2	17.9	5.3	6.0	0.35	0.20	45.6	47.4
900	18.1	18.8	6.2	6.4	0.24	0.18	48.1	49.3
1,800	18.8	20.1	6.8	6.8	0.22	0.15	49.8	51.6
3,600	20.2	20.8	7.5	7.7	0.22	0.12	53.3	55.2
Mean	18.6a	19.4a	6.4a	6.7a	0.26a	0.16b	49.2a	50.9a
Effect	L**		L**		L**		L**	
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		<0.0001	
Gypsum (P)	0.1207		0.1211		<0.0001		0.1246	
L*P	0.1399		0.1291		0.1337		0.1414	
Lime, kg ha ⁻¹	<u>depth: 0.20–0.40 m</u>							
0	12.2	12.5	5.0	5.8	0.19	0.11	36.5	37.9
900	14.0	14.2	5.4	5.8	0.15	0.11	40.3	40.9
1,800	15.2	15.9	5.6	6.1	0.15	0.07	42.3	43.6
3,600	15.9	16.0	5.6	7.0	0.13	0.08	43.4	46.4
Mean	14.3a	14.7a	5.4b	6.2a	0.16a	0.09b	40.6a	42.2a
Effect	L*		L*		L*		L*	
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		<0.0001	
Gypsum (P)	0.0927		<0.0001		<0.0001		<0.0001	
L*P	0.1054		0.1106		0.1207		0.1002	
Lime, kg ha ⁻¹	<u>depth: 0.40–0.60 m</u>							
0	7.3	11.6	3.3	4.3	0.13	0.64	22.7	33.2
900	8.2	12.0	3.8	4.4	0.10	0.32	27.5	35.9
1,800	9.6	12.8	4.0	4.7	0.10	0.18	28.7	36.9
3,600	11.2	13.0	4.1	4.9	0.10	0.13	33.0	36.9
Mean	9.1b	12.3a	3.8b	4.6a	0.1b	0.3a	28.0a	35.7b
Effect	L**		L*		L*		L*	
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		0.0315	
Gypsum (P)	<0.0001		<0.0001		<0.0001		0.0402	
L*P	0.0953		0.0845		0.1085		0.0859	

* Significant at the 0.05 probability level. ** Significant at the 0.01 probability level. ^(a) Gypsum treatments (0 and 1,700 kg ha⁻¹). ^(b) Values followed by the same letter in the rows, within each variable, are not significantly different at $P \leq 0.05$ according to the LSD test.

^(c) L represents linear effect by polynomial regression.

The binding energy of the Ca, Mg and K exchangeable cations to the soil colloids depends on the valence and the size of the hydrated ion, such that in well-drained soils, the quantities of leached K are relatively greater than those of the bivalent cations. For K, it may be observed that the application of increasing rates of limestone reduced the contents of this nutrient throughout the profile, regardless of the use of gypsum. These results may have occurred because, with the addition of limestone in a soil where the levels of Ca and Mg were already at a level considered to be high (RAIJ et al. 1996), there was displacement of K beyond the 0.6 m depth evaluated. In addition, it is possible that the plants found better growing conditions at the greater limestone application rates and, therefore, took up greater amounts of K.

Gypsum led to an increase in the Ca contents throughout the soil profile. In addition to movement of Ca, gypsum also led to displacement of exchangeable Mg up to the depth of 0.10 m, shown by the lower Mg contents found when gypsum was applied. Gypsum provided for a gradual increase in Mg up to the depth of 0.40-0.60 m, corroborating the data found by Caires et al. (2006) in wheat. Thus, although the leaching of Mg to deeper soil layers is considered to be a negative effect of the use of gypsum, this effect was not so serious for sugarcane since the sugarcane root system is deep and extensive.

For phosphorus (Table 3), there was also a linear increase in the soil contents with increasing rates of limestone application. The use of limestone assisted desorption of P because, with the rise in pH, there was an increase in the solubility of the iron and aluminum phosphates and an increase in the OH⁻ concentration in the soil solution, reducing the adsorption of phosphate in the solid phase of the soil (CASAGRANDE et al., 2003; CAMARGO et al., 2010).

The application of limestone led to a linear increase in the SO₄⁻² contents throughout the soil layers. With application of limestone, there is a rise in pH (Table 1), which promotes the increase in SO₄⁻² availability both by reducing its adsorption (CASAGRANDE et al., 2003) and by increasing release of the adsorbed amount. In addition, there was also greater mineralization of the soil organic matter with a rise in soil pH (ALVAREZ et al., 2007).

The application of gypsum significantly increased the P contents in the soil throughout the profile. In spite of the small quantity of P present in the gypsum (1%), it was able to increase the P contents by around 1 mg dm⁻³. Other authors such as Caires et al. (2011) and Silva et al. (2010) also observed an increase in phosphorus contents through the application of gypsum.

Gypsum led to an increase in the SO₄⁻² contents in all the soil layers; however, it was observed that in the deepest layers, 0.20-0.40 and 0.40-0.60 m, the greatest contents of SO₄⁻² were found, indicating descent of this nutrient in the soil profile. The small retention of S-SO₄⁻² in the soil surface layers may be attributed to the greater pH values observed in these layers. The higher pH promotes the predominance of negative electrical charges, which favor the movement of S-SO₄⁻² (SORATTO; CRUSCIOL, 2008b).

Organic matter was not affected either by the application of limestone or by the application of gypsum in the layers evaluated. The acidity amendments, through increasing pH, may accelerate mineralization of organic matter and may reduce its content in the soil in a short time. In this study, however, there was no change in the organic matter content in the soil profile through surface application of limestone and gypsum, corroborating the data of Zambrosi et al. (2007) in an area under the no-tillage system with surface application of limestone and gypsum.

Table 3. Values of P, sulfate and organic matter of soil samples taken 12 months after dolomitic lime and gypsum surface application in green sugarcane, and ANOVA significance at five soil depths in Pirassununga, São Paulo, Brazil.

Characteristic	P		S-SO ₄ ⁻²		SOM	
Gypsum, kg ha ⁻¹	0 ^(a)	1,700	0	1,700	0	1,700
	-----mg dm ⁻³ -----				g dm ⁻³	
Lime, kg ha ⁻¹	<u>depth: 0.0–0.05 m</u>					
0	3.6	5.9	5.7	8.3	28.2	28.7
900	6.8	7.6	5.9	10.0	26.5	26.4
1,800	9.3	10.1	7.2	11.9	27.1	27.9
3,600	14.1	16.2	9.4	15.0	26.8	27.4
Mean	8.4b ^(b)	9.9a	7.0b	11.3a	27.1a	27.6a

Effect	L ^{(c)**}		L*		ns	
ANOVA						
Lime (L)	<0.0001		0.0481		0.2857	
Gypsum (P)	0.0422		<0.0001		0.2694	
L*P	0.1328		0.1934		0.2677	
Lime, kg ha ⁻¹			<u>depth: 0.05–0.10 m</u>			
0	2.4	4.0	4.4	7.3	24.5	25.7
900	4.9	5.0	4.7	8.0	23.5	24.0
1,800	5.6	5.9	5.2	9.2	25.0	25.2
3,600	6.4	7.0	5.4	38.3	25.4	23.8
Mean	4.8b	5.4a	4.9b	15.7a	26.8a	27.4a
Effect	L*		L*		ns	
ANOVA						
Lime (L)	<0.0001		0.0398		0.1927	
Gypsum (P)	0.0227		<0.0001		0.1815	
L*P	0.1404		0.1158		0.2098	
Lime, kg ha ⁻¹			<u>depth: 0.10–0.20 m</u>			
0	2.4	3.0	4.1	10.0	19.4	21.8
900	3.1	4.6	4.2	10.9	22.7	22.7
1,800	4.6	4.9	5.3	12.1	22.6	23.1
3,600	5.1	6.3	6.4	40.7	22.6	23.4
Mean	3.8b	4.7a	5.0b	18.4a	21.8a	22.8a
Effect	L*		L*		ns	
ANOVA						
Lime (L)	<0.0001		0.0462		0.2011	
Gypsum (P)	0.0405		<0.0001		0.2222	
L*P	0.1221		0.1474		0.2351	
Lime, kg ha ⁻¹			<u>depth: 0.20–0.40 m</u>			
0	1.3	3.2	6.2	27.4	21.2	22.5
900	1.9	3.3	6.5	29.4	22.0	21.8
1,800	2.0	3.3	7.5	39.7	23.1	22.9
3,600	2.3	3.4	8.6	46.6	21.7	21.8
Mean	1.9b	3.3a	7.2b	35.8a	22.0a	22.2a
Effect	L*		L*		ns	
ANOVA						
Lime (L)	<0.0001		0.0478		0.2124	
Gypsum (P)	0.0209		<0.0001		0.2099	
L*P	0.1528		0.1394		0.2188	
Lime, kg ha ⁻¹			<u>depth: 0.40–0.60 m</u>			
0	1.0	2.1	12.9	40.8	20.7	21.6
900	1.3	2.6	14.3	45.4	21.0	23.0
1,800	1.7	2.7	23.2	49.7	19.8	22.0
3,600	1.9	3.7	26.3	52.2	20.6	21.1
Mean	1.5b	2.8a	19.2b	47.1a	20.5a	21.9a
Effect	L*		L*		ns	
ANOVA						
Lime (L)	<0.0001		0.0325		0.1869	
Gypsum (P)	0.0407		<0.0001		0.1992	
L*P	0.1531		0.1787		0.2025	

*Significant at the 0.05 probability level, ** significant at the 0.01 probability level, and ^{ns} represents no effect. ^(a) Gypsum treatments (0 and 1,700 kg ha⁻¹). ^(b) Values followed by the same letter in the rows, within each variable, are not significantly different at $P \leq 0.05$ according to the LSD test. ^(c) L represents linear effect by polynomial regression.

For sugar concentration, purity, fiber and reducing sugars (Table 4), there were no effects from application of limestone and gypsum. For gypsum in ratoon cane without burning, Carvalho et al. (2013) also observed a lack of response in sugarcane for such characteristics, indicating that the use of soil amendments does not hurt the quality of raw material for industry.

In contrast, for stalk and sugar yield there was a quadratic increase from lime application, both in the presence and in the absence of gypsum. It may be observed that the greatest increases occurred when the application rate recommended by Spironello et al. (1996) and double that rate was used. These results may be associated with soil acidity amendment (Table 1) and nutrient availability in the soil profile (Tables 2 and 3) brought about by the application of higher rates of

limestone. These results suggest the application of higher rates of soil acidity amendments so as to amend soil acidity at greater depths, to make nutrients available and to increase stalk and sugar yields.

Application of gypsum led to an increase in stalk and sugar yield when applied alone and at the lower lime application rate (900 kg ha⁻¹). At high rates of limestone application, an effect on stalk and sugar yield from gypsum was not observed. These results may be associated with soil acidity correction (Table 1) and the availability of nutrients in the soil profile (Tables 2 and 3) brought about by the application of high rates of limestone, thus reducing the effects of gypsum in the first year of growth after the application of these products on sugarcane.

Table 4. Values of Pol, purity, fiber, reducing sugar, and stalk, sugar, bagasse, and trash yield taken 12 months (at the time of harvest) after dolomitic lime and gypsum surface application in green sugarcane, and ANOVA significance in Pirassununga, São Paulo, Brazil.

ANOVA significance in Passandunga, São Paulo, Brazil.								
Characteristic	Pol		Purity		Fiber		Reducing sugar	
	0 ^(a)	1,700	0	1,700	0	1,700	0	1,700
Gypsum, kg ha ⁻¹								
Lime, kg ha ⁻¹	----- % -----							
0	18.2	18.3	91.8	92.1	12.8	12.5	0.40	0.40
900	18.0	17.8	92.3	91.5	12.9	12.4	0.40	0.41
1,800	17.9	17.7	92.1	91.7	12.7	12.5	0.40	0.44
3,600	18.5	17.6	92.0	91.2	12.6	12.4	0.39	0.43
Mean	18.2a ^(b)	17.8a	92.0a	91.6a	12.7a	12.4a	0.40a	0.42a
Effect	ns		ns		ns		ns	
ANOVA								
Lime (L)	0.2203		0.4423		0.3925		0.3251	
Gypsum (P)	0.2587		0.4607		0.3867		0.3349	
L*P	0.2341		0.4492		0.3922		0.2981	
	Stalk yield		Sugar yield		Bagasse Biomass ^(d)		Trash Biomass ^(e)	
Lime, kg ha ⁻¹	----- t ha ⁻¹ -----							
0	62.0b	71.4a	11.3b	13.1a	15.87b	17.85b	4.34b	5.00a
900	72.3b	79.9a	13.0b	14.2a	18.65b	19.82b	5.06b	5.59a
1,800	80.5a	80.6a	14.4a	14.3a	20.45a	20.15a	5.64a	5.64a
3,600	80.7a	82.0a	14.9a	14.4a	20.34a	20.34a	5.65a	5.74a
Mean								
Effect	Q ^{(c)*}	Q**	Q**	Q*	Q*	Q**	Q*	Q**
ANOVA								
Lime (L)	<0.0001		<0.0001		<0.0001		<0.0001	
Gypsum (P)	0.0401		0.0399		0.0322		0.0375	
L*P	0.0327		0.0426		0.0247		0.0305	

* Significant at the 0.05 probability level, ** significant at the 0.01 probability level, and ^{ns} represents no effect. ^(a) Gypsum treatments (0 and 1,700 kg ha⁻¹). ^(b) Values followed by the same letter in the rows, within each variable, are not significantly different at $P \leq 0.05$ according to the LSD test. ^(c) Q represents quadratic effect by polynomial regression. ^(d) With 50% moisture. ^(e) Trash biomass = stalk yield x 0.14 x 0.50 (considering that 1 Mg of stalk produces 140 kg of trash and that 50% of post-harvest residues were collected from the soil surface).

Furthermore, it also noted that there was small increases in stalk and sugar yield at higher rates of limestone application, associated with gypsum application or not, compared to lower rates of limestone (900 kg ha^{-1}) associated with gypsum. For that reason, at lower rates of limestone, gypsum enhanced the effect of limestone in improving chemical soil conditions (Tables 2 and 3), leading to an increase in yield (Table 4). The results of bagasse and trash production followed the same trend observed for stalk yield, indicating that higher rates of lime, in addition to promoting an increase in stalk and sugar yield, increased the amount of bagasse and trash produced. Consequently, more energy and second generation bioethanol can be generated from these sources.

CONCLUSIONS

The application of gypsum contributes so that the effects of surface application of limestone in soil acidity are able to reach the subsoil layers. However, care should be taken in application of gypsum because it leads to leaching of Mg and K from the topsoil layers.

The higher lime surface application rates used in this experiment show acidity amendment and a consequent increase in fertility in the soil profile and stalk, sugar, bagasse, and trash yield in sugarcane.

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RESUMO: O objetivo do presente estudo foi avaliar a correção da fertilidade do solo e produtividade de colmos, açúcar, bagaço e palhço a partir da aplicação superficial de calcário dolomítico e gesso em cana soca. O arranjo fatorial foi composto por 4 doses de calcário dolomítico (0, 900, 1800 e 3600 kg ha^{-1}) e duas doses de gesso (0 e 1700 kg ha^{-1}). O experimento foi realizado no período de outubro de 2006 a outubro de 2007, iniciando-se na estação chuvosa após a colheita da segunda soqueira. A variedade utilizada foi a SP813250 em plantio comercial de cana crua da Usina de São Luiz ($47^{\circ}25'33'' \text{W}$; $21^{\circ}59'46'' \text{S}$, 627 M de altitude), localizado em Pirassununga, estado de São Paulo, no sudeste do Brasil. Após 12 meses, o experimento foi colhido, determinado os parâmetros tecnológicos e amostras de solo foram coletadas e analisadas. O gesso atua como um condicionador subsuperficial contribuindo assim, para que os benefícios da aplicação superficial de calcário na correção da acidez do solo, alcancem camadas mais profundas e, portanto, permitindo o desenvolvimento do sistema radicular das plantas em maior profundidade. No entanto, a aplicação de gesso pode proporcionar lixiviação de Mg e K das camadas superficiais. As maiores doses de calcário aplicadas superficialmente propiciaram correção da acidez do solo e consequente incremento na fertilidade do solo, produtividade de colmos, açúcar, bagaço e palhço da cana.

PALAVRAS CHAVES: *Sacharum spp.* Atributos químicos do solo. Colmos. Açúcar. Bagaço e palhço.

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